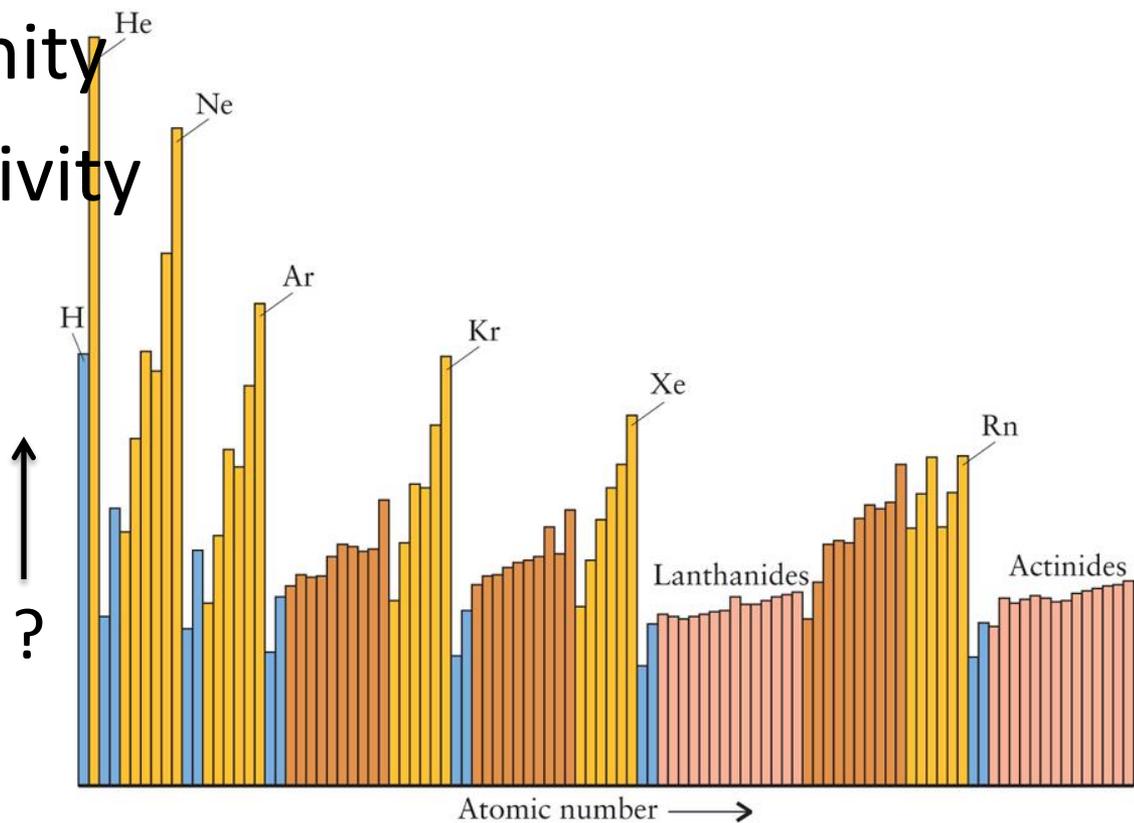


# Which of the following might represent the term on the y-axis?

1. Atomic radius
2. Ionization energy
3. Electron affinity
4. Electronegativity
5. 1 or 2
6. 2, 3, or 4



# Which of the following might represent the term on the y-axis?

7% 1. Atomic radius

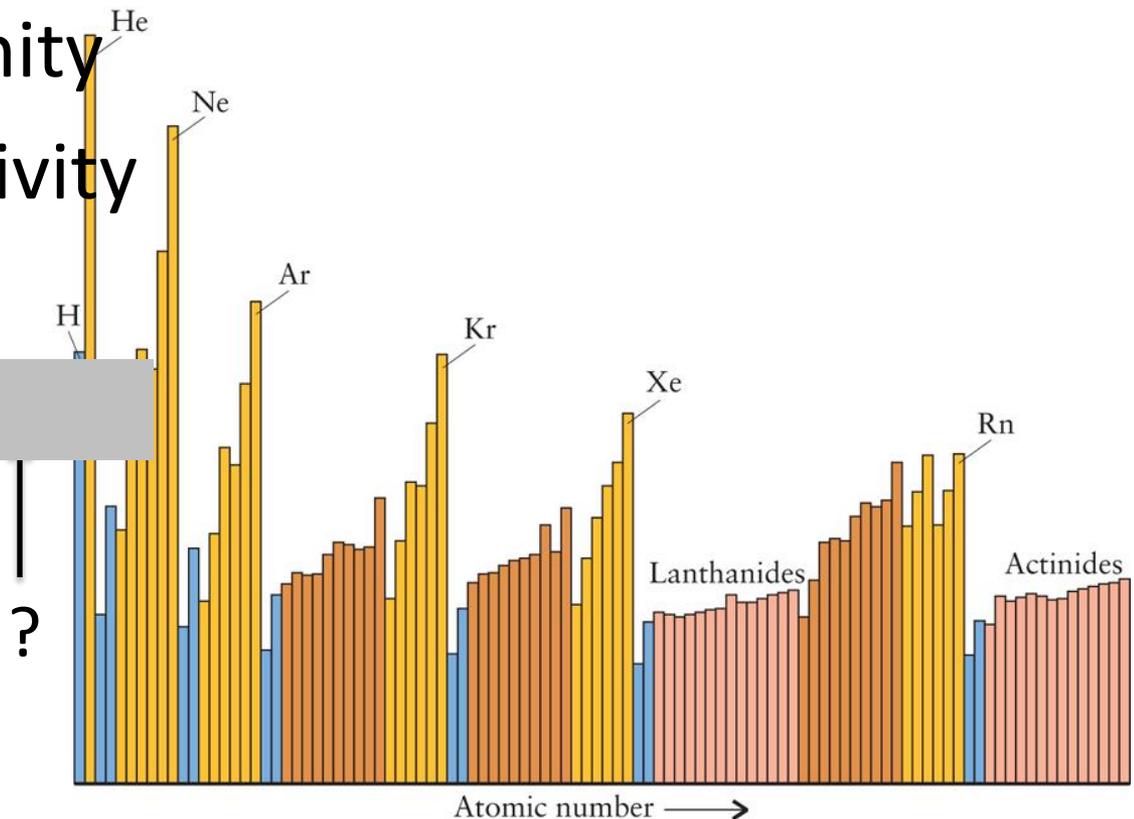
53% 2. 😊 Ionization energy

3% 3. Electron affinity

3% 4. Electronegativity

7% 5. 1 or 2

28% 6. 2, 3, or 4



# Which is correct?

1. Struct #1 Struct #2

$$FC_{OA} = 0 \quad FC_{OA} = 0$$

$$FC_{OB} = +1 \quad FC_{OB} = +1$$

$$FC_{OC} = -1 \quad FC_{OC} = -1$$

2. Struct #1 Struct #2

$$FC_{OA} = 0 \quad FC_{OA} = -1$$

$$FC_{OB} = +1 \quad FC_{OB} = +1$$

$$FC_{OC} = -1 \quad FC_{OC} = 0$$

3. Struct #1 Struct #2

$$FC_{OA} = -2 \quad FC_{OA} = -2$$

$$FC_{OB} = 0 \quad FC_{OB} = 0$$

$$FC_{OC} = -2 \quad FC_{OC} = -2$$

4. Struct #1 Struct #2

$$FC_{OA} = 0 \quad FC_{OA} = 1$$

$$FC_{OB} = -1 \quad FC_{OB} = -1$$

$$FC_{OC} = 1 \quad FC_{OC} = 0$$

1.

2.

3.

4.

# Which is correct?

1. Struct #1 Struct #2

$$FC_{OA} = 0 \quad FC_{OA} = 0$$

$$FC_{OB} = +1 \quad FC_{OB} = +1$$

$$FC_{OC} = -1 \quad FC_{OC} = -1$$

3. Struct #1 Struct #2

$$FC_{OA} = -2 \quad FC_{OA} = -2$$

$$FC_{OB} = 0 \quad FC_{OB} = 0$$

$$FC_{OC} = -2 \quad FC_{OC} = -2$$

 2. Struct #1 Struct #2

$$FC_{OA} = 0 \quad FC_{OA} = -1$$

$$FC_{OB} = +1 \quad FC_{OB} = +1$$

$$FC_{OC} = -1 \quad FC_{OC} = 0$$

4. Struct #1 Struct #2

$$FC_{OA} = 0 \quad FC_{OA} = 1$$

$$FC_{OB} = -1 \quad FC_{OB} = -1$$

$$FC_{OC} = 1 \quad FC_{OC} = 0$$

10% 1.

69% 2.

9% 3.

12% 4.



# Which molecule is nitric oxide?

85%  1. NO

8% 2. N<sub>2</sub>O

7% 3. HNO<sub>2</sub>

1s						1s			
H						He			
2s-filling				2p-filling					
Li	Be			B	C	N	O	F	Ne
3s-filling				3p-filling					
Na	Mg			Al	Si	P	S	Cl	Ar
4s-filling		3d-filling				4p-filling			



# Determine the FC for the doubled-bonded F atom in our $\text{BF}_3$ Lewis Structure

74% 😊 1. +1

14% 2. +2

5% 3. 0

5% 4. -1

2% 5. -2

1s								1s					
H								He					
2s-filling								2p-filling					
Li	Bc							B	C	N	O	F	Ne
3s-filling								3p-filling					
Na	Mg							Al	Si	P	S	Cl	Ar
4s-filling		3d-filling						4p-filling					

How many **additional** resonance structures are there for  $\text{CrO}_4^{2-}$ ?

1. One
2. Two
3. Three
4. Four
5. Five
6. Six
7. Seven
8. Eight
9. Zero

How many **additional** resonance structures are there for  $\text{CrO}_4^{2-}$ ?

3%

1. One

8%

2. Two

6%

3. Three

70%



4. Four

3%

5. Five

7%

6. Six

0%

7. Seven

1%

8. Eight

3%

9. Zero

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